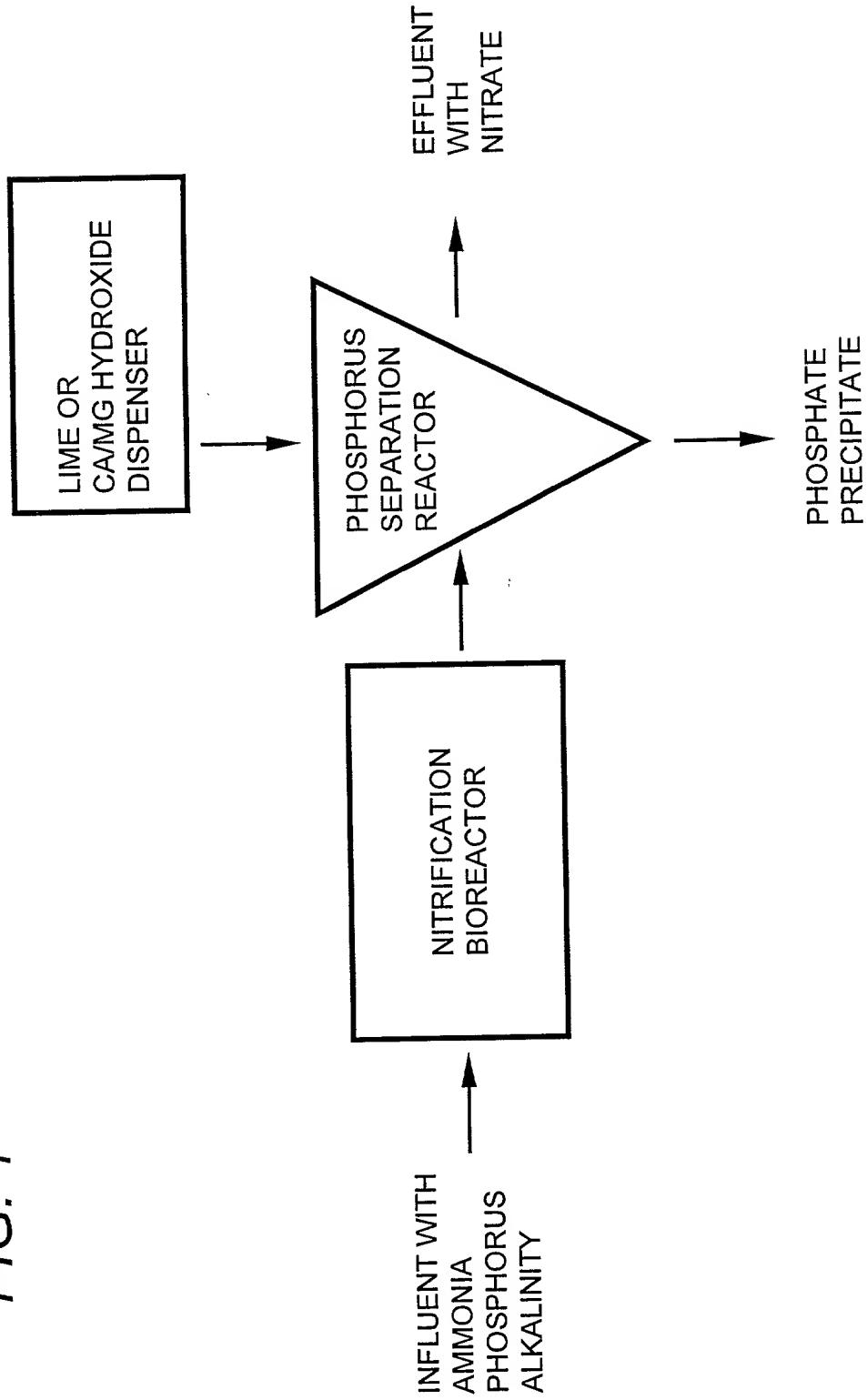


FIG. 1



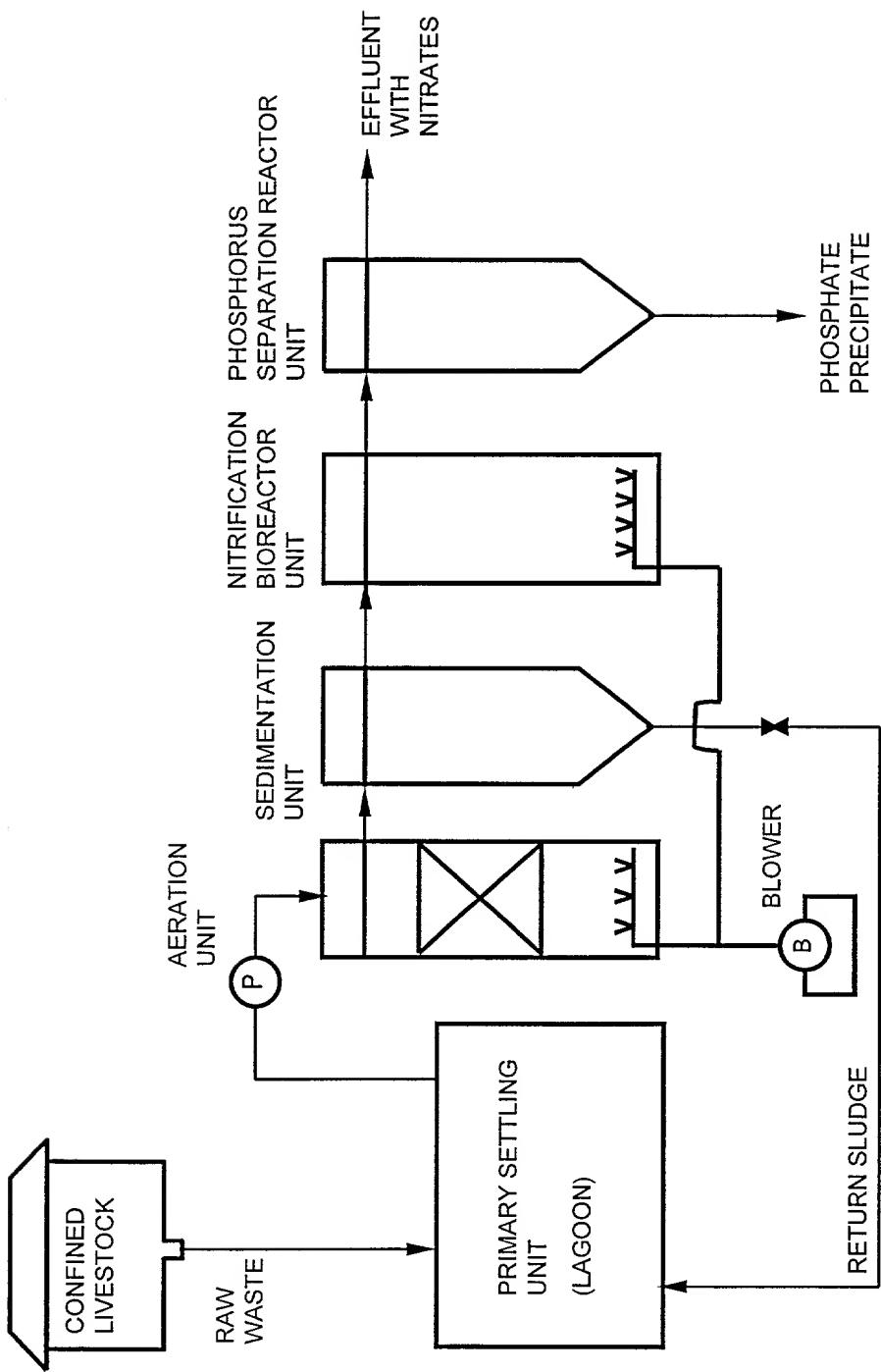
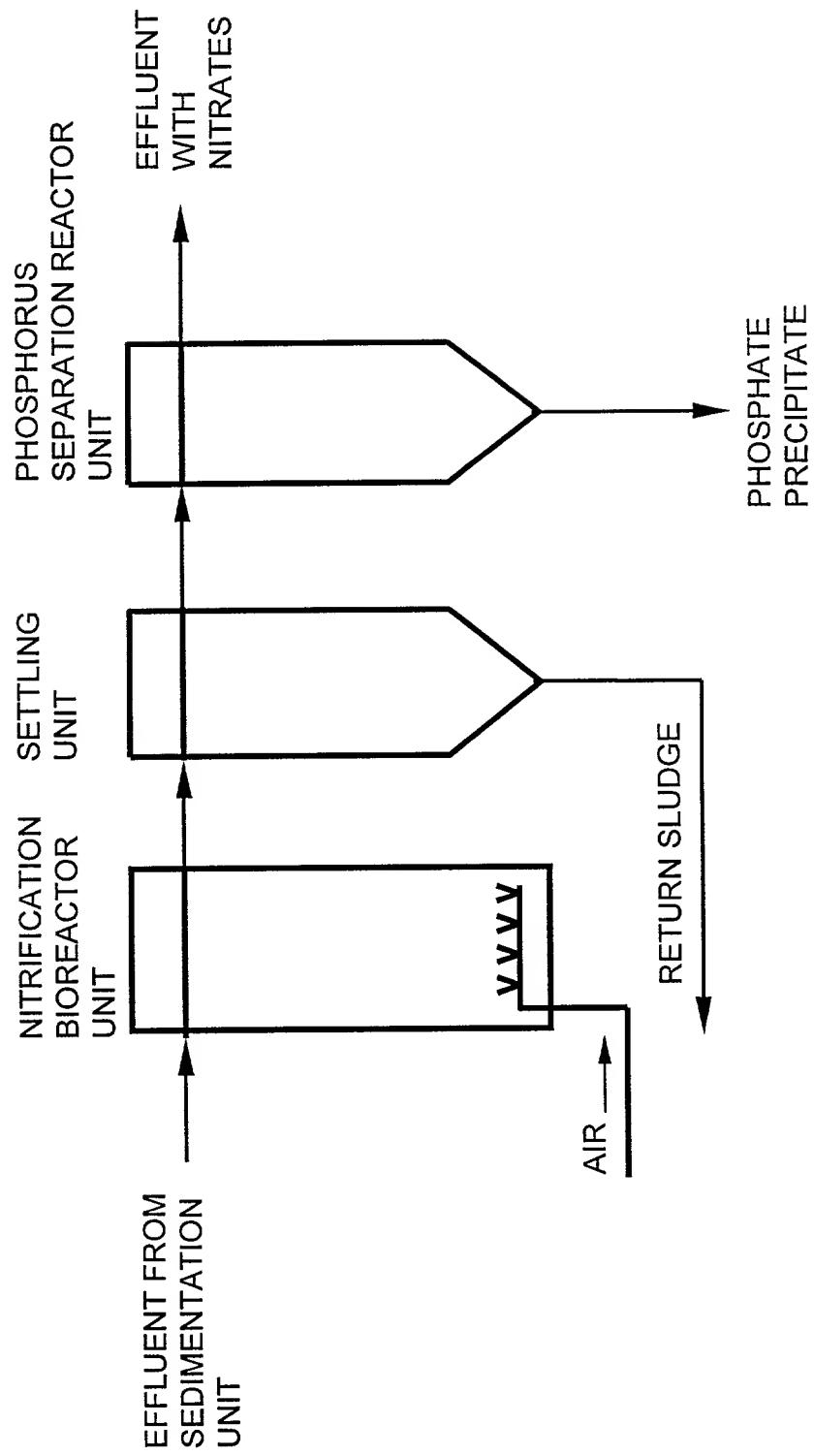
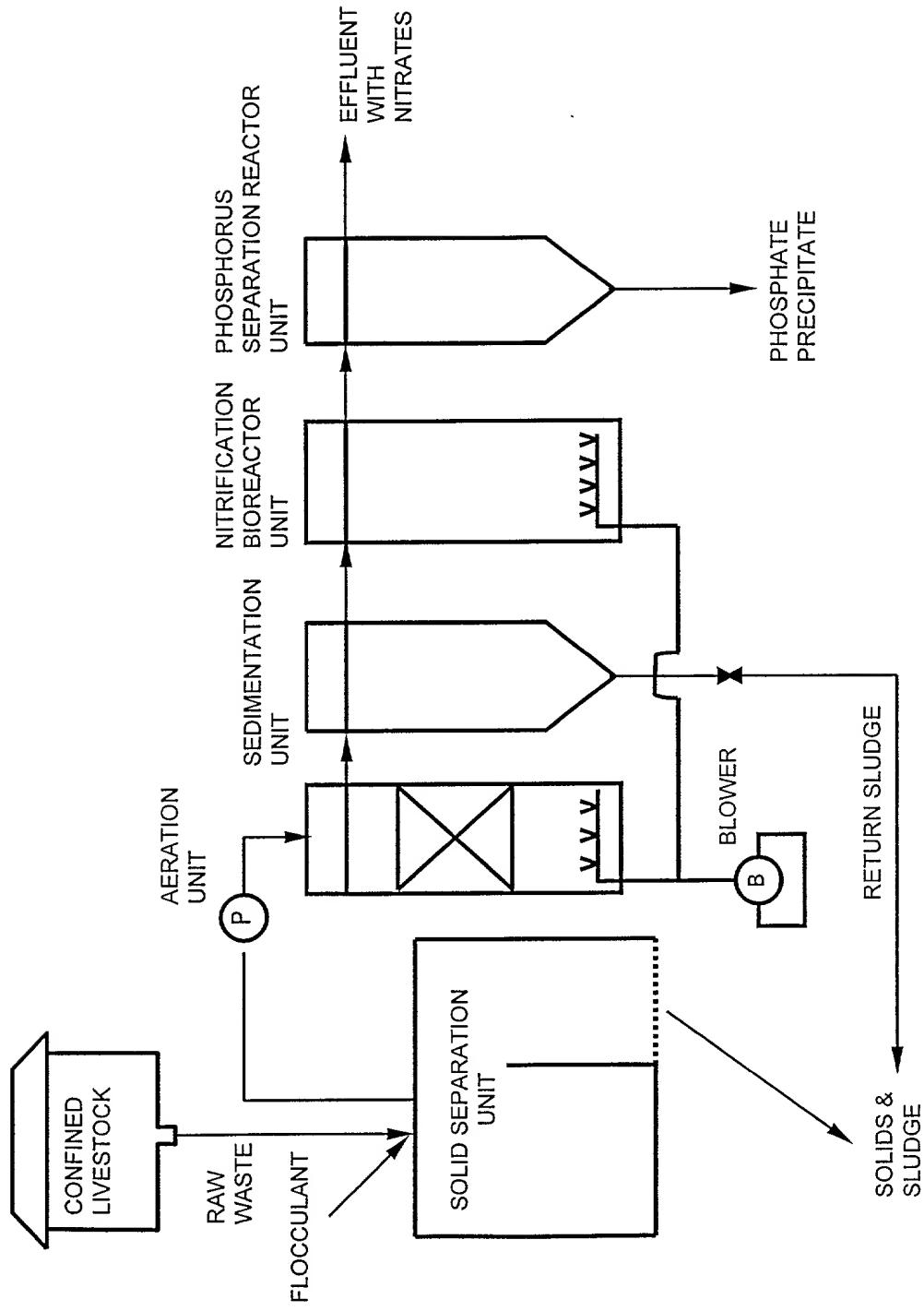


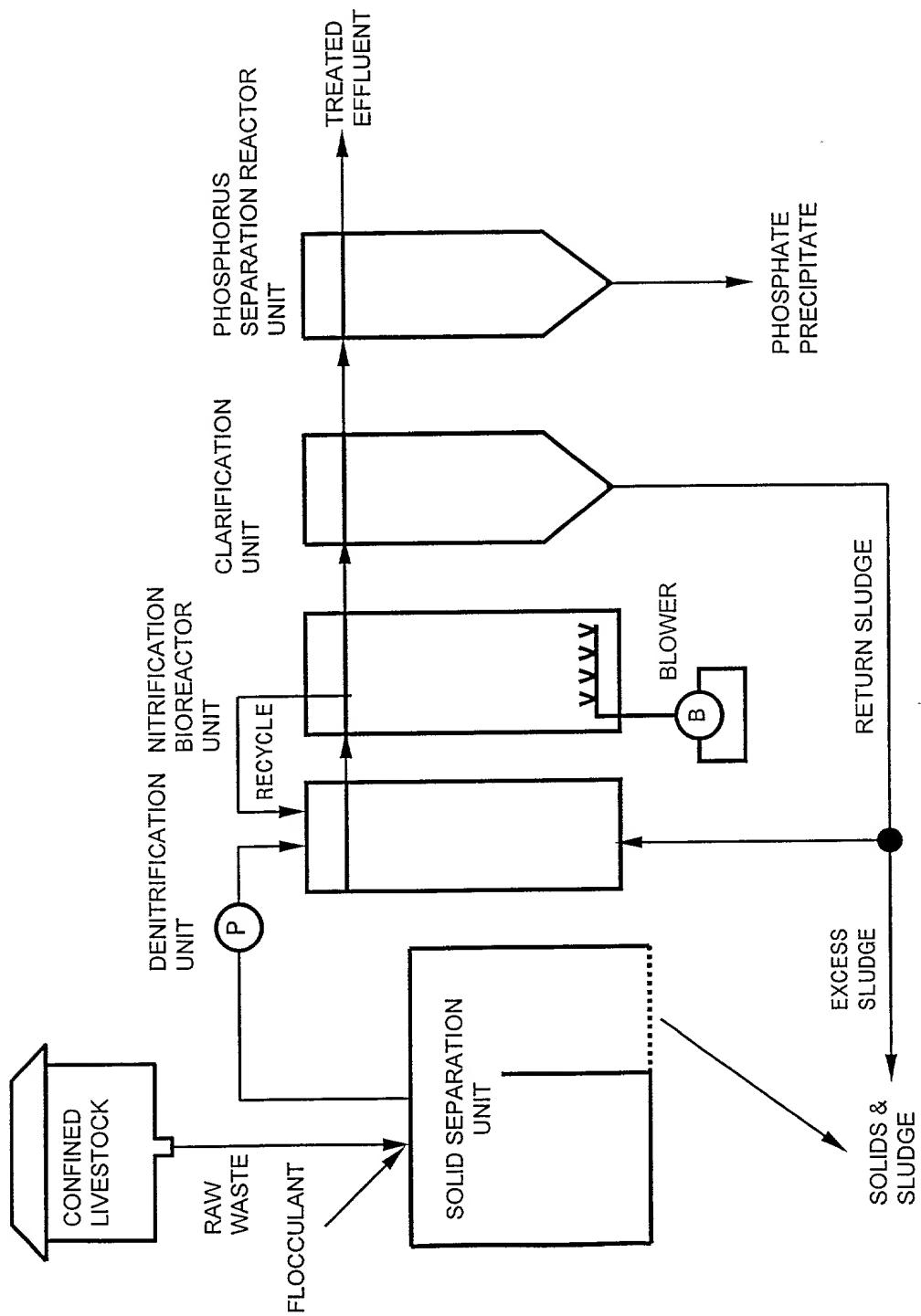
FIG. 2

*FIG. 3*



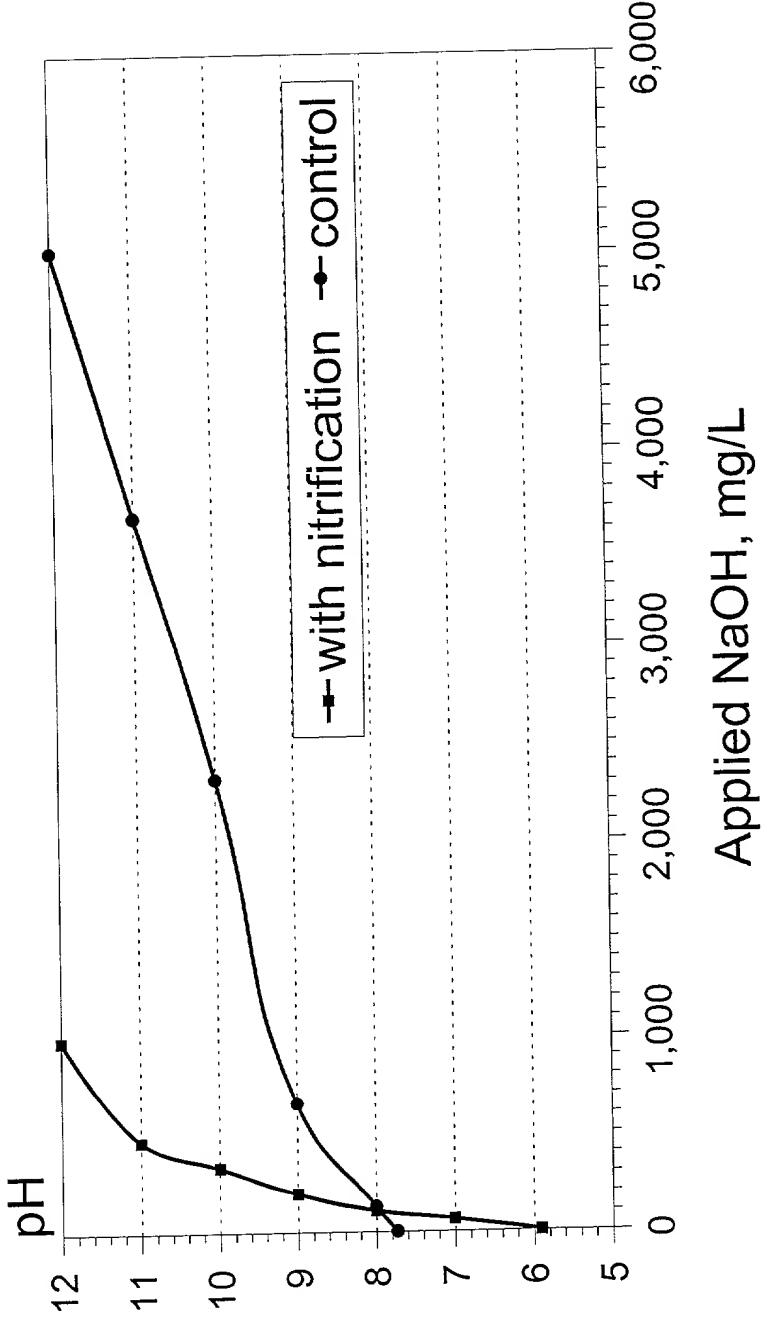


**FIG. 4**



*FIG. 5*

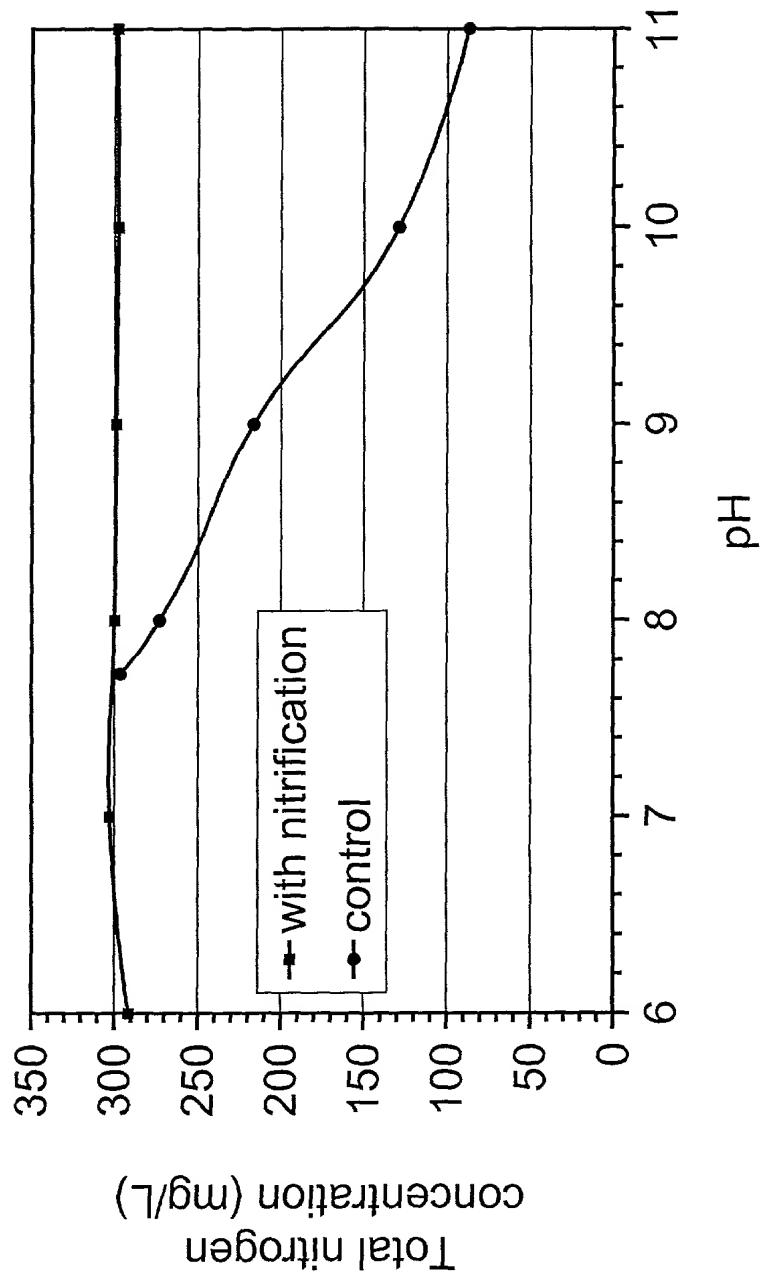
FIG. 6



EFFECT OF ALKALI ADDITION ON pH OF SWINE WASTEWATER THAT RECEIVED NITRIFICATION PRETREATMENT VS. CONTROL

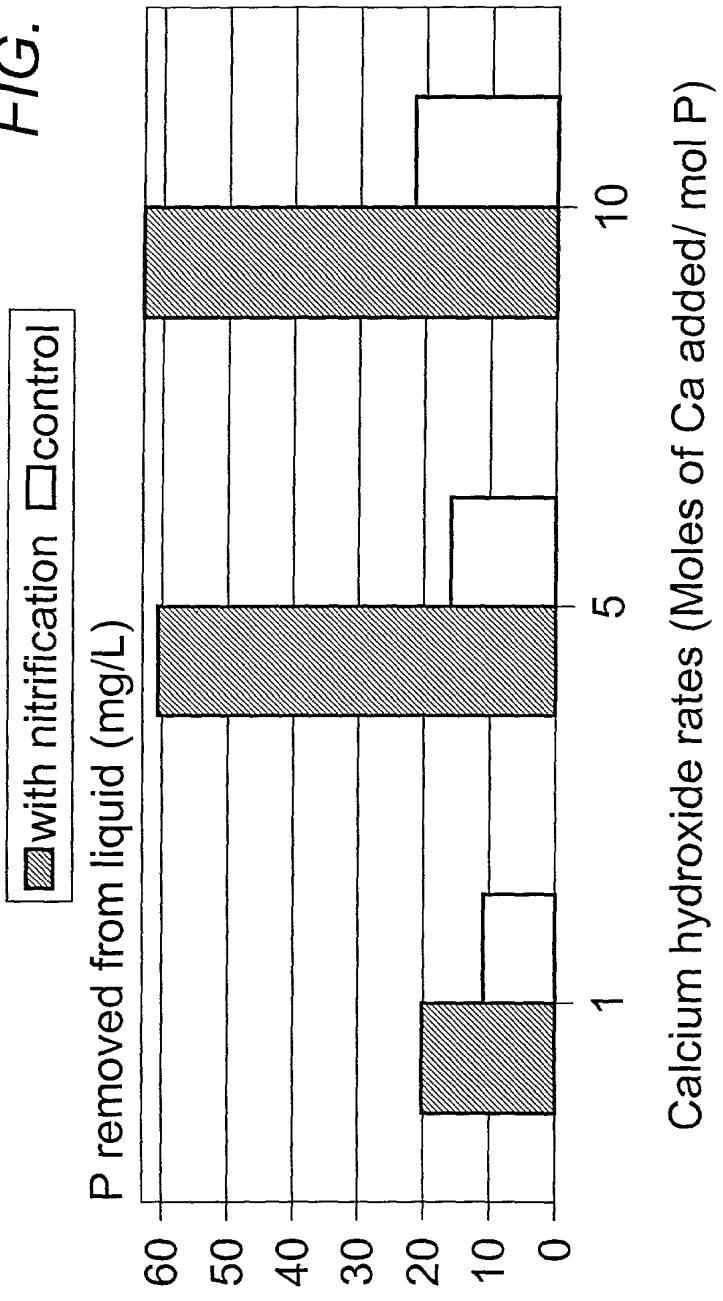
Nitrogen losses in swine wastewater  
by ammonia volatilization

**FIG. 7**



Phosphorus removal from swine wastewater  
using Calcium Hydroxide

FIG. 8



Initial conditions:

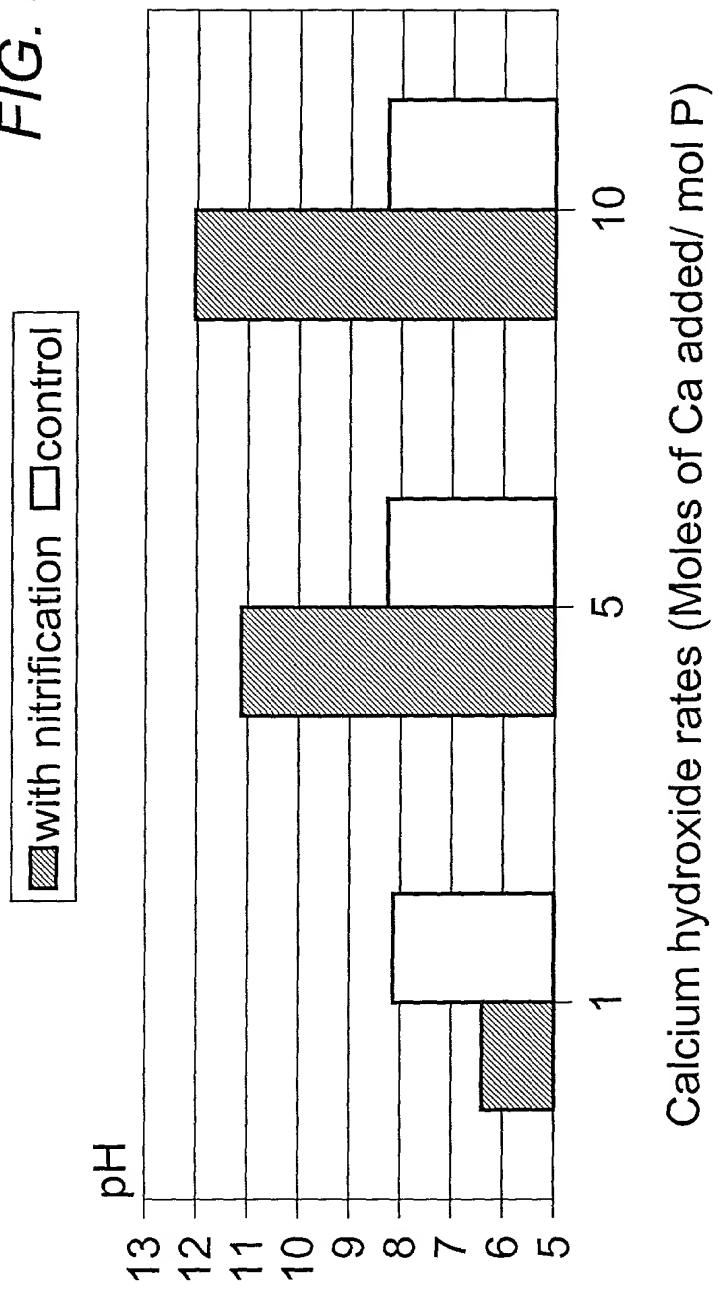
$\text{PO}_4\text{-P} = 63 \text{ mg/L}$ ,  $\text{pH} = 8.05$ , alkalinity =  $1890 \text{ mg/L}$ ,  $\text{NH}_4\text{-N} = 300 \text{ mg/L}$

After nitrification:

$\text{PO}_4\text{-P} = 63 \text{ mg/L}$ ,  $\text{pH} = 6.06$ , alkalinity =  $63 \text{ mg/L}$ ,  $\text{NH}_4\text{-N} = 61 \text{ mg/L}$

Phosphorus removal from swine wastewater  
using Calcium Hydroxide: effect on pH

FIG. 9

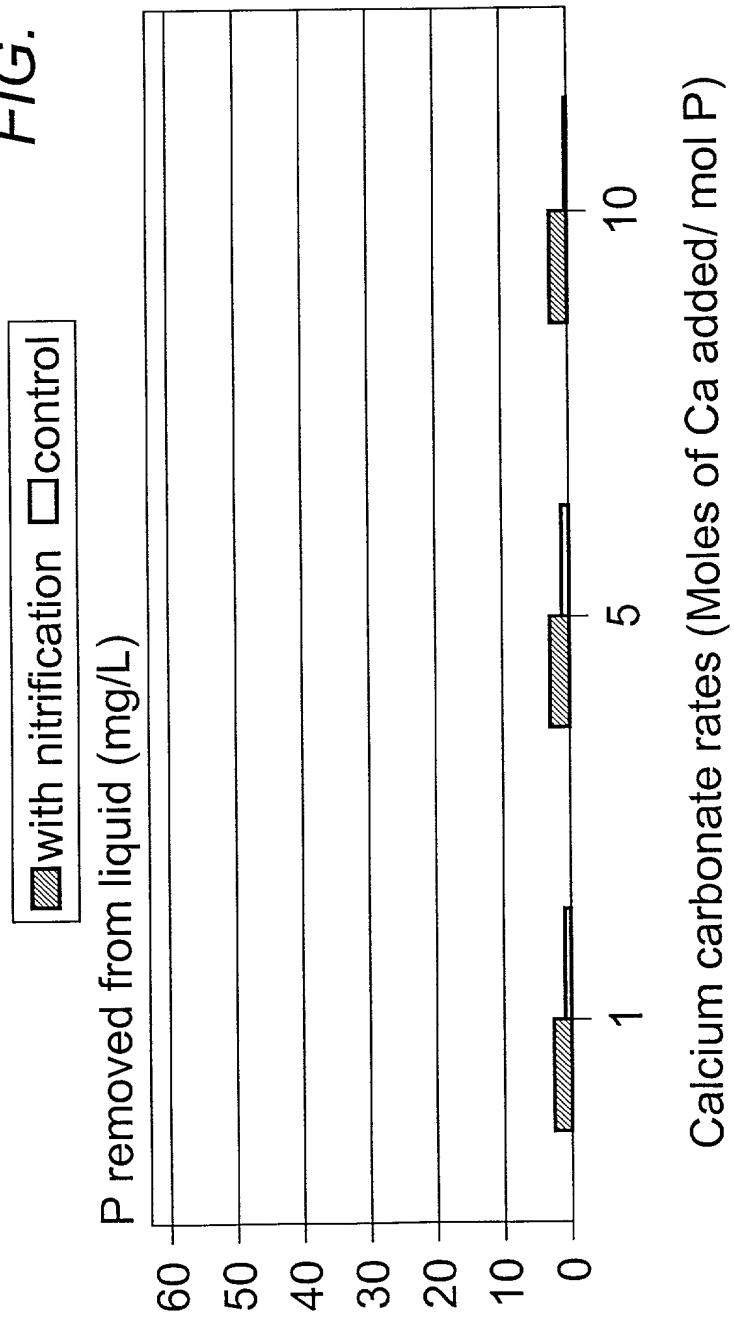


Initial conditions:  
 $\text{PO}_4\text{-P} = 63 \text{ mg/L}$ ,  $\text{pH} = 8.05$ , alkalinity =  $1890 \text{ mg/L}$ ,  $\text{NH}_4\text{-N} = 300 \text{ mg/L}$

After nitrification:  
 $\text{PO}_4\text{-P} = 63 \text{ mg/L}$ ,  $\text{pH} = 6.06$ , alkalinity =  $63 \text{ mg/L}$ ,  $\text{NH}_4\text{-N} = 61 \text{ mg/L}$

Use of Calcium Carbonate Lime was not effective  
for removal of phosphorus from swine wastewater

FIG. 10



Initial conditions:

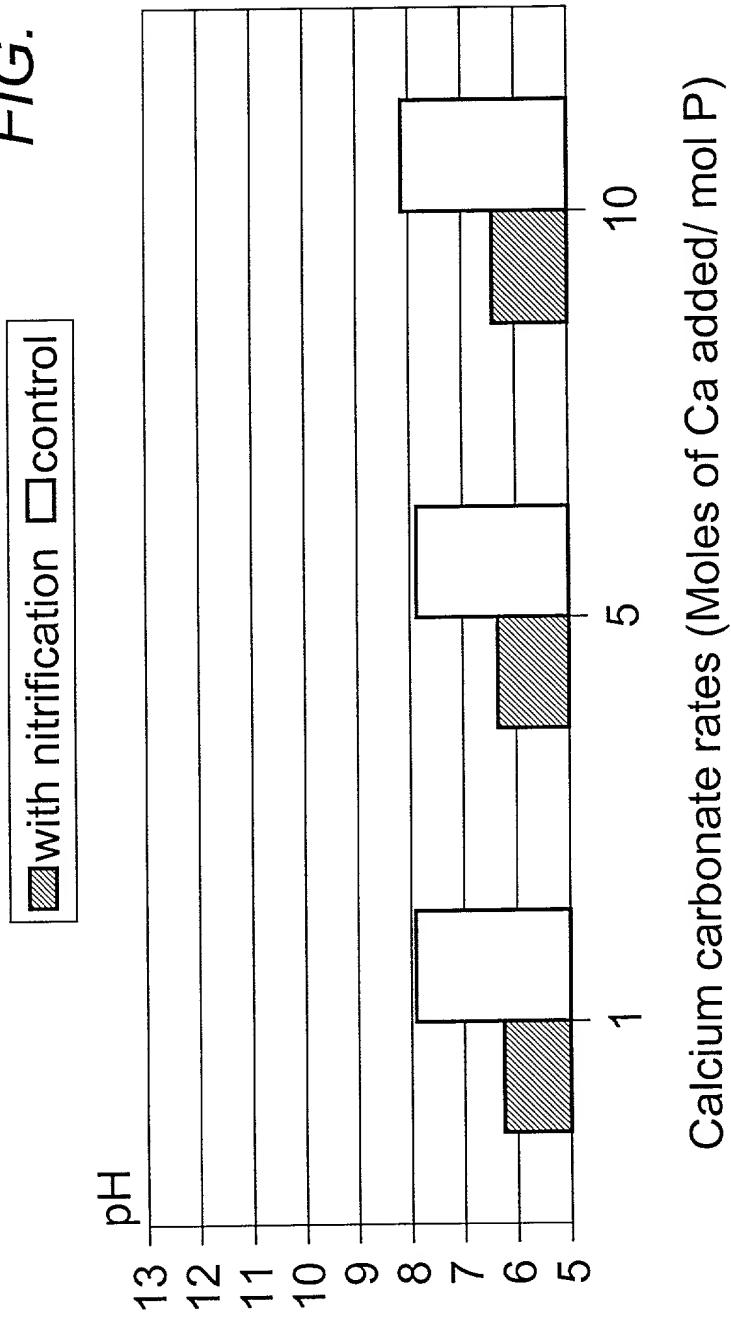
$\text{PO}_4\text{-P} = 63 \text{ mg/L}$ ,  $\text{pH} = 8.05$ , alkalinity = 1890 mg/L,  $\text{NH}_4\text{-N} = 300 \text{ mg/L}$

After nitrification:

$\text{PO}_4\text{-P} = 63 \text{ mg/L}$ ,  $\text{pH} = 6.06$ , alkalinity = 63 mg/L,  $\text{NH}_4\text{-N} = 61 \text{ mg/L}$

Application of Carbonate lime to swine wastewater  
did not affect pH or phosphorus removal.

FIG. 11



Initial conditions:

$\text{PO}_4\text{-P} = 63 \text{ mg/L}$ ,  $\text{pH} = 8.05$ , alkalinity = 1890 mg/L,  $\text{NH}_4\text{-N} = 300 \text{ mg/L}$

After nitrification:

$\text{PO}_4\text{-P} = 63 \text{ mg/L}$ ,  $\text{pH} = 6.06$ , alkalinity = 63 mg/L,  $\text{NH}_4\text{-N} = 61 \text{ mg/L}$